

Spectroscopic phase microscopy for quantifying hemoglobin concentrations in intact red blood cells

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We report a practical method for label-free quantification of specific molecules using spectroscopic imaging of sample-induced phase shifts. Diffraction phase microscopy equipped with various wavelengths of light source is used to record wavelength-dependent phase images. We first perform dispersion measurements on pure solutions of single molecular species present in the cells, such as albumin and hemoglobin (Hb). With this prior calibration of molecular specific dispersion, we demonstrate the extraction of Hb concentration from individual human red blood cells. The end point of this study is noninvasive monitoring of physiological states of intact living cells. © 2009 Optical Society of America
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Biochemical activity of molecules governs cellular metabolism, which makes molecular concentrations important indicators of pathophysiological states of cells or tissues. Staining agents such as fluorescent dye and fluorescent protein have been widely used to quantify molecules of interest. However, the quantification of molecules by such techniques requires careful consideration of the artifacts of the staining agents: photobleaching, interference with metabolic activities of living cells, and nonuniform binding to the target molecules.

The refractive index (RI) can provide molecular concentration without such artifacts. As is well known, the RI is related to the concentration of cellular dry mass [1]. Quantitative phase microscopy has proved to be a useful tool to measure the phase delay induced by a specimen, which is related to its RI [2–12]. However, two limitations prevent it from retrieving specific molecular concentrations from the phase-delay measurements. First, the measured phase delay implies approximate total protein contents, not the concentration [1,13]. The phase delay is an integrated RI in the cell volume, and the RI should be decoupled from thickness to obtain molecular concentration. Second, both the phase and the RI are not specific enough to differentiate single molecular species from mixtures. All types of molecule present in the cell can contribute to the RI. We note that the wavelength-dependent RI, obtained via dispersion, can resolve both limitations simultaneously. Many types of molecule can be differentiated by their dispersion. For example, hemoglobin (Hb) has significant dispersion in visible wavelengths. Dispersion was previously used to decouple RI from the thickness of cells from the dispersion of a fluorescent medium [7] and to measure the concentration of molecules in solution and in tissue sections [14].

This Letter presents a new technique, spectroscopic phase microscopy (SPM) that can measure the concentration of specific molecules in living cells and their volumes at the same time. SPM integrates diffraction phase microscopy (DPM) [5,15] and a white-light source with various bandpass filters to select various wavelengths. We demonstrate the simultaneous extraction of Hb concentration and cell volume of intact individual red blood cells (RBCs). Abnormalities in Hb concentration are associated with various RBC diseases (sickle-cell anemia, thalassemia, and malaria infection) [6], and the present technique will potentially lead to assessing these disease conditions quantitatively.

The experimental setup consists of two parts: a white-light source with color filters for wavelength selection, and DPM for measuring quantitative phase delay (Fig. 1). DPM provides highly stable phase measurements, while the use of color filters simplifies the experimental setup in switching the wave-

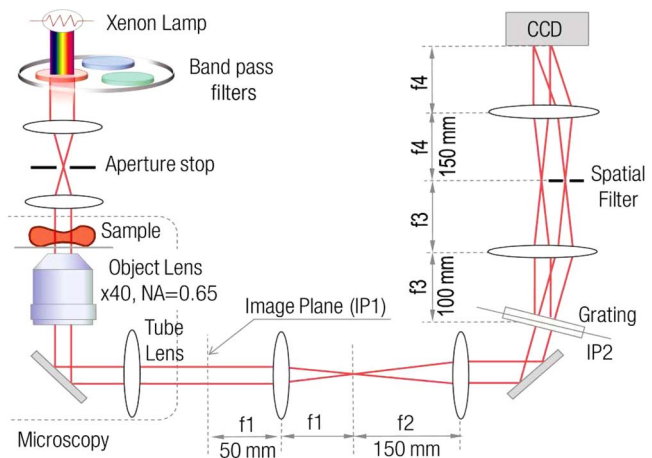


Fig. 1. (Color online) Experimental setup.

length of the light source. In dispersion measurements, a wide range of wavelengths is necessary. Either a tunable laser or a set of different lasers can be used, either of which is costly. Instead, we used a xenon arc lamp with a set of bandpass filters. Seven different filters are used to select various center wavelengths: 440 ± 20 , 546 ± 10 , 560 ± 20 , 580 ± 25 , 600 ± 20 , 655 ± 20 , and 700 ± 20 nm. The common-path geometry of DPM matches the optical path lengths for the sample and reference arms such that the alignment is independent of the wavelength and temporal coherence of the illumination source. However, it still requires long spatial coherence, because of a spatial filtering process to be described. Thus, the size of aperture stop was minimized to increase spatial coherence. A specimen, located at the sample stage of an inverted microscope, is projected onto image plane IP1. The image was further magnified and delivered to IP2, where a holographic grating was placed to generate multiple diffraction orders. The zeroth- and first-order beams were isolated. The zeroth-order beam is spatially low-pass filtered by using a pinhole in a $4f$ lens system; the beam then became a clean plane wave at the camera plane. The first-order beam served as sample beam. Both beams interfered and generated a spatially modulated interference image, which was then captured by a CCD camera. Since both beams share almost the same beam path, common-mode phase noise was canceled out on interference. The electric field was extracted from the recorded interferogram by a spatial Hilbert transform [5]. The grating period, $30 \mu\text{m}$, was set to be smaller than the diffraction-limited spot of the microscopic imaging system at the grating plane. All the lenses were achromatic to minimize chromatic dispersion.

To assess the ability of SPM, we measured the RIs of polydimethylsiloxane (PDMS), bovine serum albumin (BSA), and Hb solutions at various wavelengths. Since the phase image presents relative phase delay between a sample and a medium, phase reference is required in order to determine the absolute phase change induced by a specimen. We used a microfluidic channel and the RI of water for this purpose. We filled the PDMS channel with distilled water and measured quantitative phase images at different wavelengths. The relative phase delay between PDMS and water is given by $\Delta\varphi = 2\pi/\lambda(n_{\text{PDMS}}(\lambda) - n_w(\lambda))h$, where λ , h , n_{PDMS} , and n_w are the wavelength of the light, channel height, RI of PDMS, and RI of water, respectively. Given the known dispersion of water [16] and $h = 10 \mu\text{m}$, we obtained n_{PDMS} [Fig. 2(g)]. Next, we measured the RI of the BSA solution. The measured dispersion of PDMS served as reference this time. The PDMS channel was filled with BSA solution (300 mg/ml; Gibco). From the relative phase to PDMS, the dispersion of BSA solutions were acquired by using the relation $\Delta\varphi = 2\pi(n_{\text{PDMS}}(\lambda) - n_{\text{BSA}}(\lambda; C))h/\lambda$, with $n_{\text{BSA}}(\lambda; C)$ the RI of BSA solution at wavelength λ and concentration C . The obtained n_{BSA} [Fig. 2(g)] is consistent with the literature [17]. Next, we measured the dispersion of Hb solutions, prepared from Hb protein powder (H7379,

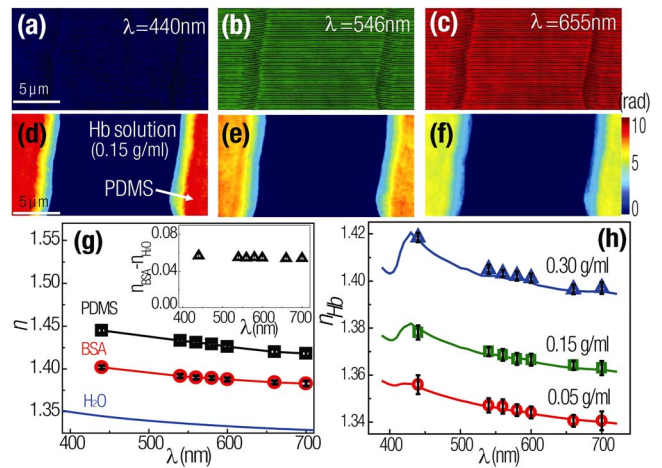


Fig. 2. (Color online) (a)–(c) Interferograms of PDMS channel filled with Hb solution. (d)–(f) Corresponding quantitative phase maps. (g) RIs of PDMS, BSA, and water. RIs of water are adopted from [16]. Inset, RI difference between BSA solution and water. (h) Measured dispersion of Hb solutions (symbols) and comparison with [18] (lines). Error bars indicate standard deviations.

Sigma-Aldrich, Inc.), at three different concentrations: 0.05, 0.15, and 0.30 g/ml, respectively [Figs. 2(a)–2(f)]. Using the method above, we obtained $n_{\text{Hb}}(\lambda; C)$ [Fig. 2(h)], which is consistent with that of the literature [18]. From this measurement, we retrieved a linear relationship between the RI of Hb solution and its concentration at various wavelengths [1]: $n_{\text{Hb}}(\lambda; C) = \alpha(\lambda)C + n_w$, with $\alpha(\lambda)$ a specific RI increment for Hb: $\alpha(440) = 0.240 \pm 0.007 \text{ ml/g}$, $\alpha(560) = 0.227 \pm 0.004 \text{ ml/g}$, and $\alpha(660) = 0.221 \pm 0.005 \text{ ml/g}$. The albumin protein in BSA solution does not show significant dispersion in the visible range [inset, Fig. 2(g)] $\alpha(\lambda)$ is $0.183 \pm 0.003 \text{ ml/g}$ for the wavelength range of 440–700 nm.

Next, using the calibrated dispersion of Hb solution, we applied SPM to measure cytoplasmic Hb concentration in live human RBCs. Fresh blood (5 ml) obtained by venipuncture were diluted in phosphate-buffered saline (PBS) solution and then washed three times to remove white blood cells and platelets. The interference images of RBCs were measured at the seven wavelengths [Figs. 3(a)–3(c)]. The phase images of an RBC can be expressed as $\Delta\varphi(x, y; \lambda) = 2\pi(\alpha(\lambda)C(x, y) + n_X(x, y, \lambda))h(x, y)/\lambda$, with $h(x, y)$ the local thickness of the cell and $n_X(x, y, \lambda)$ an relative averaged RI of other molecules besides Hb in RBC compared with water (non-Hb proteins (2.5% of total protein mass) and ionic molecules in cytosolic space [19]). These non-Hb molecules usually do not have distinct dispersion at visible wavelengths. For example, albumin, the most abundant plasma protein in mammals, has almost the same trend in RI as water in the wavelength range 440–700 nm [inset, Fig. 2(g)]. Thus, we can approximate $n_X(x, y, \lambda)$ as an independent constant of wavelength, n_X . Since $\alpha(\lambda)$ was calibrated above, there are three unknowns: C , n_X , and $h(x, y)$. Thus, the phase measurements at three different wavelengths are sufficient to extract C , n_X , and $h(x, y)$ simultaneously. In principle, we can

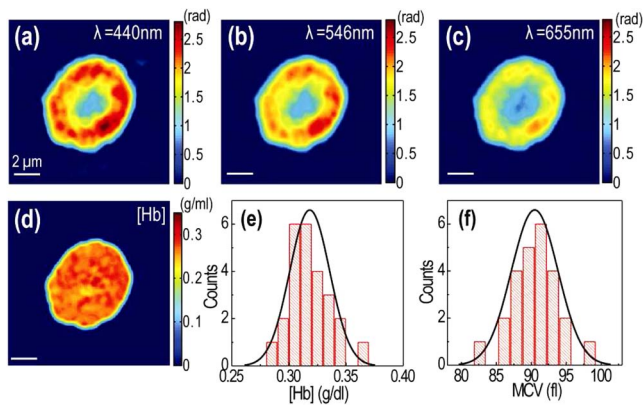


Fig. 3. (Color online) (a)–(c) Quantitative phase maps of an RBC at three different wavelengths. (d) Retrieved Hb concentration. Histogram of (e) Hb concentration and (f) mean cell volumes ($N=25$).

distinguish N different molecular contents with $N+2$ spectroscopic phase measurements, assuming that they have distinctive dispersion from one another. Figure 3(d) shows the retrieved Hb concentration, $C(x,y)$ of an RBC. Figure 3(e) is a histogram of the Hb concentration in RBCs ($N=25$). The average value is 0.318 ± 1.7 g/ml. From the measured area of the cells and corresponding thickness, we calculated the cell volume [Fig. 3(f)]. The average value is 90.5 ± 3.3 fl. These results are within the normal physiological range [19]. According to the measured conversion factors, the sensitivity of SPM is 0.0009 – 0.0014 g/ml for Hb concentration, since the sensitivity in phase measurement of DPM is ~ 3 mrad [15]. This sensitivity corresponds to 0.08 – 0.13 pg for Hb contents per cell, assuming the average red blood cell volume is 90 fl.

In summary, we have presented SPM, a simple and practical method for spectroscopic phase imaging, which can simultaneously measure Hb concentrations and cell volume of living RBCs. The information provided is quantitative. In the instrument, illumination is accomplished by means of a white-light source typically delivered via a microscopy. Thus incorporating the instrument to a commercial microscopy should be straightforward, which makes our instrument readily available to many biology laboratories. As demonstrated, dispersion provides molecular specificity; imaging the quantitative phase maps at different wavelengths can help differentiate among molecules. This new instrument has many potential applications, one of which we give here; quantifying

isomeric shifts in Hb (oxy- and deoxy-Hb) when exposed to different oxygen pressures. Binding of oxygen to Hb proteins cause conformational changes that result in significant differences in dispersion. Thus, an SPM instrument may provide a valuable tool for better understanding of oxygen transport by RBCs.

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References

1. R. Barer, *Nature* **172**, 1097 (1953).
2. E. Cuche, F. Bevilacqua, and C. Depeursinge, *Opt. Lett.* **24**, 291 (1999).
3. A. Barty, K. Nugent, D. Paganin, and A. Roberts, *Opt. Lett.* **23**, 817 (1998).
4. C. G. Rylander, D. P. Davé, T. Akkin, T. E. Milner, K. R. Diller, and A. J. Welch, *Opt. Lett.* **29**, 1509 (2004).
5. G. Popescu, T. Ikeda, R. R. Dasari, and M. S. Feld, *Opt. Lett.* **31**, 775 (2006).
6. Y.-K. Park, M. Diez-Silva, G. Popescu, G. Lykotrafitis, W. Choi, M. S. Feld, and S. Suresh, *Proc. Natl. Acad. Sci. USA* **105**, 13730 (2008).
7. B. Rappaz, F. Charriere, C. Depeursinge, P. J. Magistretti, and P. Marquet, *Opt. Lett.* **33**, 744 (2008).
8. Y. Park, W. Choi, Z. Yaqoob, R. Dasari, K. Badizadegan, and M. Feld, *Opt. Express* **17**, 12285 (2009).
9. B. Kemper and G. von Bally, *Appl. Opt.* **47**, 52 (2008).
10. B. Rappaz, M. Marquet, E. Cuche, Y. Emery, C. Depeursinge, and P. J. Magistretti, *Opt. Express* **13**, 9361 (2005).
11. G. Popescu, Y. Park, W. Choi, R. Dasari, M. Feld, and K. Badizadegan, *Blood Cells Mol. Dis.* **41**, 10 (2008).
12. N. Shaked, M. Reinhart, and A. Wax, *Opt. Lett.* **34**, 767 (2009).
13. G. Popescu, Y. Park, N. Lue, C. Best-Popescu, L. Deflores, R. Dasari, M. Feld, and K. Badizadegan, *Am. J. Physiol.: Cell Physiol.* **295**, C538 (2008).
14. C. Yang, A. Wax, I. Georgakoudi, E. Hanlon, K. Badizadegan, R. Dasari, and M. Feld, *Opt. Lett.* **25**, 1526 (2000).
15. Y. K. Park, G. Popescu, K. Badizadegan, R. R. Dasari, and M. S. Feld, *Opt. Express* **14**, 8263 (2006).
16. P. Schiebener, J. Straub, J. Sengers, and J. Gallagher, *J. Phys. Chem. Ref. Data* **19**, 677 (1990).
17. M. Andersen, L. Painter, and S. Nir, *Biopolymers* **13**, 1261 (1974).
18. M. Friebel and M. Meinke, *Appl. Opt.* **45**, 2838 (2006).
19. M. Lichtman, *Williams Hematology* (McGraw-Hill Professional, 2005).