

Absorption and Circular Dichroism Spectra of Molecular Aggregates With the Full Cumulant Expansion

Lorenzo Cupellini,* Filippo Lipparini, and Jianshu Cao


 Cite This: *J. Phys. Chem. B* 2020, 124, 8610–8617


 Read Online

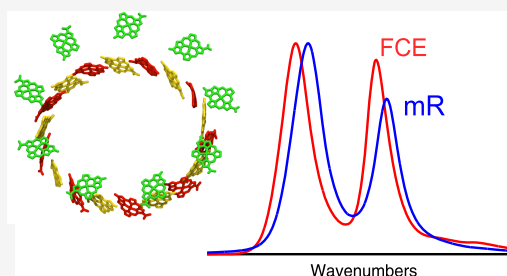
ACCESS |


 Metrics & More


 Article Recommendations


 Supporting Information

ABSTRACT: The exciton Hamiltonian of multichromophoric aggregates can be probed by spectroscopic techniques such as linear absorption and circular dichroism. To compare calculated Hamiltonians to experiments, a lineshape theory is needed, which takes into account the coupling of the excitons with inter- and intramolecular vibrations. This coupling is normally introduced in a perturbative way through the cumulant expansion formalism and further approximated by assuming a Markovian exciton dynamics, for example with the modified Redfield theory. Here, we present the implementation of the full cumulant expansion (FCE) formalism (*J. Chem. Phys.* 142, 2015, 094106) to efficiently compute absorption and circular dichroism spectra of molecular aggregates beyond the Markov approximation, without restrictions on the form of exciton–phonon coupling. By employing the LH2 system of purple bacteria as a challenging test case, we compare the FCE lineshapes with the Markovian lineshapes obtained with the modified Redfield theory, showing that the latter presents a less satisfying agreement with experiments. The FCE approach instead accurately describes the lineshapes, especially in the vibronic sideband of the B800 peak. We envision that the FCE approach will become a valuable tool for accurately comparing model exciton Hamiltonians with optical spectroscopy experiments.



By employing the LH2 system of purple bacteria as a challenging test case, we compare the FCE lineshapes with the Markovian lineshapes obtained with the modified Redfield theory, showing that the latter presents a less satisfying agreement with experiments. The FCE approach instead accurately describes the lineshapes, especially in the vibronic sideband of the B800 peak. We envision that the FCE approach will become a valuable tool for accurately comparing model exciton Hamiltonians with optical spectroscopy experiments.

1. INTRODUCTION

The optical and photophysical properties of chromophore aggregates such as light-harvesting complexes are determined by the interactions between the monomer excitations and by the coupling between these excitations and the vibrational degrees of freedom.^{1,2} These interactions are quantified, respectively, by the exciton couplings between transitions localized on individual chromophores and by the spectral density of the exciton–phonon coupling. Together, these interactions determine the excited-state dynamics of light-harvesting systems.

Excitonic and vibronic couplings can be parameterized by fitting the linear and nonlinear spectra of the aggregates, guided by prior knowledge of their structure.^{3–6} Detailed high-resolution structures of chromophoric aggregates have also enabled the use of quantum chemistry (QM) methods to compute a priori both excitonic Hamiltonians and vibronic couplings.^{1,7–14} However, these quantities still need to be compared with experiments, by simulating the aggregate spectra.

While the spectrum of a single chromophore can be evaluated exactly starting from the corresponding spectral density,¹⁵ the mixing of electronic states resulting from the exciton couplings prevents a simple calculation of the overall spectrum without further approximation. In the delocalized basis of electronic eigenstates, the exciton–phonon coupling is not diagonal anymore; the off-diagonal exciton–phonon coupling cannot be treated exactly. Numerically exact methods, such as the hierarchical equations of motion (HEOM)^{16–19} and the stochastic path integral (sPI),²⁰ are computationally very expensive and limited to benchmark calculations on model

systems. On the other hand, perturbative approaches based on Redfield equations have been extensively used,^{12,13,21,22} but their applicability range is limited.²³ The main concern about the Redfield-like approaches arises from the Markovian and secular approximations; in particular, the Markovian approximation is known to break down when vibrational modes are in resonance with electronic energy gaps.²³ Several strategies have been devised to overcome the Markov/secular approximations.^{23–32} Among these, the full second-order cumulant expansion (FCE) offers the most general form of non-Markov/secular approach within perturbative treatment of the off-diagonal exciton–phonon coupling.²⁸ For linear absorption properties, the FCE is equivalent to the second-order perturbative time-convolutionless (TCL2) quantum master equation.²⁸ Despite the approximation to the real non-Markovian dynamics, the TCL2 method performs well for linear absorption spectra, although it is qualitatively inaccurate for nonlinear spectra.³³

In this contribution, we implement the FCE approach to compute absorption and circular dichroism (CD) lineshapes of molecular aggregates, without restrictions on the functional form or the site-dependence of spectral densities. We show an

Received: June 8, 2020

Revised: September 3, 2020

Published: September 9, 2020



application of the method to the LH2 system of purple bacteria,^{34,35} using excitonic parameters¹⁰ and spectral densities¹³ recently calculated by some of us. We demonstrate that a Redfield-based approach suffers some limitations in describing the absorption lineshape of LH2, while the FCE approach yields an excellent agreement with the experiments, offering a new method to compare QM exciton calculations with absorption and circular dichroism spectra.

2. METHODS

2.1. Hamiltonian. The total Hamiltonian of the excitonic aggregate, $\hat{\mathcal{H}}$, can be written as $\hat{\mathcal{H}} = \hat{\mathcal{H}}_{\text{el}} + \hat{\mathcal{H}}_{\text{ph}} + \hat{\mathcal{H}}_{\text{el-ph}}$ where the electronic part $\hat{\mathcal{H}}_{\text{el}}$ is the exciton Hamiltonian

$$\hat{\mathcal{H}}_{\text{el}} = \sum_i \xi_i |i\rangle \langle i| + \sum_{ij} V_{ij} |i\rangle \langle j| \quad (1)$$

where ξ_i is the vertical excitation energy of chromophore i , and V_{ij} is the electronic coupling between the excited states of chromophores i and j . The eigenstates μ of the exciton Hamiltonian are expressed in the site basis as

$$|\mu\rangle = \sum_j c_j^\mu |j\rangle \quad (2)$$

where the excitonic coefficients $c_j^\mu = \langle j|\mu\rangle$ are the eigenvectors of the exciton matrix.

The bath correlation function C_n of chromophore n is determined by the spectral density

$$C_n(t) = \frac{1}{\pi} \int_0^\infty d\omega \left[\coth\left(\frac{\beta\hbar\omega}{2}\right) \cos(\omega t) - i \sin(\omega t) \right] \tilde{C}_n(\omega) \quad (3)$$

2.2. Absorption and Circular Dichroism. The absorption spectrum is given by the half-sided Fourier transform of the transition dipole correlation function.²⁶ In the site basis

$$A(\omega) \propto \omega \Re \int_0^\infty dt e^{i\omega t} C_{\mu\mu}(t) \\ = \omega \Re \int_0^\infty dt e^{i\omega t} \sum_{ij} \text{tr}_b[\rho_g \mu_i(t) \mu_j(0)] \quad (4)$$

where ρ_g is the ground-state density matrix of the bath and the i, j indices run on the chromophores of the system. Applying the Condon approximation, the transition dipoles can be taken out of the bath trace

$$A(\omega) \propto \omega \Re \int_0^\infty dt e^{i\omega t} \sum_{ij} \mathbf{M}_{ij} \mathbf{I}_{ij}(t) \quad (5)$$

where $\mathbf{I}(t)$ is the absorption tensor of the aggregate²⁸

$$\mathbf{I}(t) = \text{tr}_b[e^{-iHt} \rho_g e^{iHt}] \quad (6)$$

and $\mathbf{M}_{ij} = \boldsymbol{\mu}_i \cdot \boldsymbol{\mu}_j$ is the dipole strength matrix.

In the same way, the CD spectrum is related to the electric dipole–magnetic dipole correlation function²⁶

$$\text{CD}(\omega) \propto \omega \Im \int_0^\infty dt e^{i\omega t} C_{\mu-m}(t) \\ = \omega \Im \int_0^\infty dt e^{i\omega t} \sum_{ij} \text{tr}_b[\rho_g \mu_i(t) \mathbf{m}_j(0)] \quad (7)$$

$$\text{CD}(\omega) \propto \omega \Im \int_0^\infty dt e^{i\omega t} \sum_{ij} (\boldsymbol{\mu}_i \cdot \mathbf{m}_j) \mathbf{I}_{ij}(t) \quad (8)$$

where \Im denotes the imaginary part. Neglecting the intrinsic magnetic moment of each chromophore, \mathbf{m}_j can be written as

$$\mathbf{m}_j = \frac{i}{2\hbar c} \mathcal{E}_j \mathbf{R}_j \times \boldsymbol{\mu}_j \quad (9)$$

where \mathcal{E}_j and \mathbf{R}_j are the excitation energy and the position, respectively, of chromophore j . One can then define the rotatory strength matrix²⁶

$$r_{ij} = \mathcal{E}_j \mathbf{R}_j (\boldsymbol{\mu}_j \times \boldsymbol{\mu}_i) \quad (10)$$

and express the CD spectrum as

$$\text{CD}(\omega) \propto \omega \Re \int_0^\infty dt e^{i\omega t} \sum_{ij} r_{ij} \mathbf{I}_{ij}(t) \quad (11)$$

Both the absorption and the CD lineshapes are determined by the absorption tensor $\mathbf{I}(t)$, which will be evaluated using FCE. The lineshape tensor can be transformed into the exciton basis as

$$\mathbf{I}_{\mu\nu}(t) = \sum_{ij} (c_i^\mu)^* c_j^\nu \mathbf{I}_{ij}(t) \quad (12)$$

where c_i^μ are the coefficients of the exciton states. In the same way, the dipole strength matrix and rotatory strength matrix can be transformed into an exciton basis. However, the matrix r_{ij} given in eq 10 is nonsymmetric and gauge-dependent. A gauge-invariant expression can be obtained by substituting the excitation energy \mathcal{E}_j with the geometric average $\sqrt{\mathcal{E}_i \mathcal{E}_j}$ and symmetrizing the matrix

$$r_{ij} = -\sqrt{\mathcal{E}_i \mathcal{E}_j} (\mathbf{R}_j - \mathbf{R}_i) (\boldsymbol{\mu}_i \times \boldsymbol{\mu}_j) \quad (13)$$

2.3. Full Cumulant Expansion for Arbitrary Spectral Densities. The absorption tensor $\mathbf{I}(t)$ can be written in the exciton basis via the full second-order cumulant expansion given in ref 28 as

$$\mathbf{I}(t) = e^{-iH_0 t} e^{-\mathbf{K}(t)} \quad (14)$$

where the FCE lineshape matrix \mathbf{K} is defined as

$$K_{\mu\nu}(t) = \sum_\alpha \sum_n X_n^{\mu\alpha} X_n^{\alpha\nu} \int_0^t dt_2 \int_0^{t_2} dt_1 e^{i\omega_\mu t_2 - i\omega_\nu t_1} C_n(t_2 - t_1) \\ = \sum_\alpha \sum_n X_n^{\mu\alpha} X_n^{\alpha\nu} F_{\mu\alpha\nu}(t) \quad (15)$$

Here, n denotes a site; μ, ν , and α refer to exciton states; $X_n^{\mu\alpha} = (c_n^\mu)^* c_n^\alpha$; $\hbar\omega_{\mu\alpha} = E_\mu - E_\alpha$; and $C_n(t)$ is the autocorrelation function of the energy gap.

The numerical integration in eq 15 can be made easier by separating the case $\omega_{\mu\nu} = 0$. First, we evaluate $F_{\mu\alpha\nu}(t)$ by integrating by parts

$$F_{\mu\alpha\nu}(t) = \frac{1}{i\omega_{\mu\nu}} \int_0^t (e^{i\omega_\mu t} - e^{i\omega_\nu t_2}) e^{i\omega_\alpha t_2} C_n(t_2) dt_2 \quad (16)$$

In the case where $\omega_{\mu\nu} = 0$, we cannot separate the integral as done below in eq 18, but noting that

$$\lim_{\omega_{\mu\nu} \rightarrow 0} \frac{(e^{i\omega_\mu t} - e^{i\omega_\nu t_2})}{i\omega_{\mu\nu}} = t - t_2$$

we reformulate eq 16 as ($\mu = \nu$)

$$F_{\mu\mu\nu}(t) = tG_{\mu\nu}(t) - H_{\mu\nu}(t) \quad (17)$$

where we have defined the auxiliary tensors $G_{\nu\alpha n}(t) = \int_0^t e^{i\omega_{\nu\alpha}t_2} C_n(t_2) dt_2$ and $H_{\mu\nu}(t) = \int_0^t t_2 e^{i\omega_{\mu\nu}t_2} C_n(t_2) dt_2$. Next, we consider the case of $\omega_{\mu\nu} \neq 0$. In this case, the integral can be separated into two terms

$$F_{\mu\alpha\nu}(t) = \frac{e^{i\omega_{\mu\nu}t}}{i\omega_{\mu\nu}} G_{\nu\alpha n}(t) - \frac{1}{i\omega_{\mu\nu}} H_{\mu\nu}(t) \quad (18)$$

2.4. Numerical Implementation. The calculation of $K_{\mu\nu}(t)$ is the computationally most intensive part of the FCE evaluation, especially when the number of distinct spectral densities increases. At the beginning of the calculation, exciton weights $W_{\mu\alpha n} = X_n^{\mu\alpha} X_n^{\alpha\nu}$ are computed and summed over the chromophores that have the same autocorrelation function $C_k(t)$

$$W_{\mu\alpha k} = \sum_{n \in k} X_n^{\mu\alpha} X_n^{\alpha\nu}$$

At each time step, only the auxiliary tensors $G_{\mu\alpha k}$ and $H_{\mu\alpha k}$ are propagated from their previous values. In this implementation, we use the trapezoid rule

$$G_{\nu\alpha k}(t + \delta t) \simeq G_{\nu\alpha k}(t) + \frac{\delta t}{2} (e^{i\omega_{\nu\alpha}t} C_n(t) + e^{i\omega_{\nu\alpha}(t+\delta t)} C_n(t + \delta t)) \quad (19)$$

$$H_{\nu\alpha k}(t + \delta t) \simeq H_{\nu\alpha k}(t) + \frac{\delta t}{2} (te^{i\omega_{\nu\alpha}t} C_n(t) + e^{i\omega_{\nu\alpha}(t+\delta t)} (t + \delta t) C_n(t + \delta t)) \quad (20)$$

The two tensors are allocated as $N_{\text{site}}^2 \times N_{\text{SD}}$, where N_{SD} is the number of unique spectral densities in the calculation. The tensor $F_{\mu\alpha\nu}$ is computed with eq 18 or 17 when $\omega_{\mu\nu}$ is smaller than a threshold.

Finally, the lineshape matrix is computed, by summing over the N_{SD} different spectral densities, giving

$$K_{\mu\nu}(t) = \sum_{\alpha} \sum_k W_{\mu\alpha k} F_{\mu\alpha k}(t)$$

The numerical calculations described in this Section were implemented in a Fortran 90 code available for download under the LGPL license agreement.³⁶

3. RESULTS

The LH2 system of purple bacteria^{34,35} provides an ideal test case for the FCE approach. LH2 comprises 27 bacteriochlorophyll a (BChl) pigments, which give rise to a typical absorption spectrum in the near infrared. The BChls in LH2 are organized into two rings, responsible for the absorption peaks at 800 nm (B800 ring) and at \sim 850 nm (B850 ring), respectively. The B850 ring contains nine $\alpha\beta$ dimers of tightly packed BChls, whose strong nearest-neighbor couplings delocalize the excitation and shift the peak frequency to the red. Conversely, the BChls in the B800 ring are more separated and weakly coupled, though the effect of this coupling on the band shape is not negligible.^{3,13} LH2 has been extensively investigated theoretically, with both modeling studies^{3,6,37–41} and quantum chemical calculations,^{10,13,42–44} and experimentally, by a variety of spectroscopic techniques.^{45–50}

To accurately compare the FCE lineshapes to experiments on LH2, we need reliable parameters for the exciton and exciton–phonon couplings. Some of us have employed a polarizable multiscale quantum chemical model to accurately compute the site energies and excitonic couplings for LH2.¹⁰ We have shown that excitonic parameters computed on the crystal structure (@CRY) represent the exciton structure of LH2 at very low temperature, whereas the parameters computed along a molecular dynamics simulation (@MD) represent LH2 at room temperature. In this work, we will employ the two sets of parameters from ref 10 to describe the spectral lineshapes at different temperatures.

We model the exciton–phonon couplings of the BChls in LH2 with a different spectral density for each chlorophyll type. We take the spectral densities calculated in ref 13 for the three different BChls in LH2. These are comprised of a high-frequency part, directly calculated through a normal-mode analysis of the BChl in the protein, and a low-frequency part, modeled as a Drude oscillator

$$C_j(\omega) = 2\lambda_{c,j} \frac{\omega\gamma_{c,j}}{\omega^2 + \gamma_{c,j}^2} + \sum_{k=0}^M S_{k,j} \omega_{k,j} \frac{\omega\gamma_{k,j}}{(\omega - \omega_{k,j})^2 + \gamma_d^2} \quad (21)$$

where j denotes the chromophore, and k denotes the vibrational modes of the chromophore, λ_c is the low-frequency contribution to the reorganization energy, and γ_c is the damping parameter of the overdamped Brownian oscillator. In ref 13, parameters λ_c and γ_c and the width σ of the static disorder were empirically adapted. The effect of static disorder was modeled by averaging the spectra over N realizations of the disordered exciton Hamiltonian, obtained by randomly choosing the site energies from a Gaussian distribution with standard deviation σ . The number N of realizations was deemed sufficient when the average spectrum of $N/2$ realizations was indistinguishable from the average of the remaining $N/2$. We thus used 5000 realizations for the 77 K spectra and 500 realizations for the 300 K spectra. The details of spectral densities and static disorder are reported in the Supporting Information.

3.1. Absorption Spectra. We first compare the LH2 absorption spectra with those obtained in ref 13 employing the modified Redfield (mR) equation. To do this, we compare in Figure 1 the spectra obtained at 77 K from the same exciton Hamiltonian (@CRY) and disorder parameters. A noticeable shift is present, for both absorption bands, between the two predicted spectra due to the neglect of the off-diagonal reorganization energy in the mR theory. In particular, the mR equation neglects the imaginary part of the off-diagonal exciton–phonon coupling.²³ As already found in ref 23 for a model system, this off-diagonal reorganization shift can be different for different exciton states. In our case, the reorganization energy is different in the two bands because (i) we explicitly used different spectral densities for the B850 and B800 BChls, and (ii) the excitation is much more delocalized in the B850 ring than in the B800 ring. The difference in reorganization energies is reflected in the apparent energy gap between the B800 and B850 peaks, which is \sim 60 cm^{-1} narrower in the FCE spectrum.

In Figure 2, we separately show the B850 and B800 band shapes, comparing the mR theory and FCE with the experiment.⁵ We shifted and normalized all spectra on the experimental maxima to facilitate the shape comparison. Apart from the reorganization energy shift, the shape of the B850 band

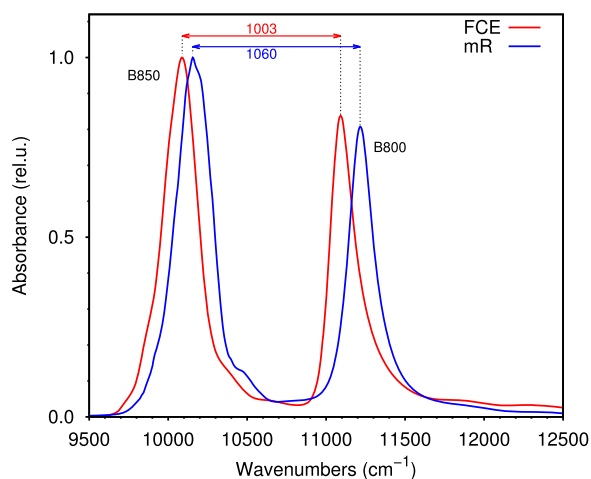


Figure 1. Absorption spectrum of LH2 at $T = 77$ K, obtained using FCE (red) and modified Redfield theory (blue), with the @CRY calculated exciton parameters (see text). The arrows indicate the energy gap in cm^{-1} between the two absorption maxima.

as predicted by mR theory is very similar to the one predicted by FCE, pointing to a small influence of non-Markovian effects in this band. In comparison, mR theory predicts a substantially symmetric shape for the B800 band, whereas the FCE band shape is strongly asymmetric. The origin of this asymmetry has already been debated and assigned to dipole strength redistribution due to exciton interactions and static disorder.^{3,37} Here, using a more realistic description of exciton and exciton–phonon couplings, we find that non-Markovian effects, which are absent in the mR description but included in the FCE, enhance the asymmetry of the B800 band.

To better show the non-Markovian effects in the B850 and B800 bands, we reported in Figure S1 the absorption spectra from two sample realizations of the static disorder, computed with FCE and mR theory. The stronger static disorder of the B850 band gives rise to lineshapes with several narrow peaks, due to the splitting and mixing of the lowest exciton levels. These narrow peaks are similarly described by mR theory and FCE. On the other hand, the B800 band is still more asymmetrical with

FCE than with mR theory, confirming the conclusions drawn from the averaged spectra.

As for room-temperature spectra, we find an even larger difference between the mR and FCE lineshapes (Figure 3). In addition to the frequency shift noticed above, the FCE spectra present a narrower B800 band, with a pronounced vibronic structure. By contrast, the mR lineshapes show noticeable Lorentzian broadening at both the red and blue tails of the spectrum, due to the interexciton relaxation rates that are overestimated by the mR theory.

In the right panel of Figure 3, we present the same comparison, changing the width σ of static disorder for the B850 BChls, and the magnitude of the low-frequency component of the spectral density, λ_c (the B800 static disorder was kept constant). Now, the B850 lineshapes are more similar, but the mR spectrum still presents Lorentzian tails. Overall, the non-Markovian effects seem much stronger for room-temperature spectra, and the mR results strongly deviate from the FCE ones.

Due to the large difference between FCE and mR spectra, for simplicity, we compare only the FCE spectra with the experiments. For this comparison, we use the disorder parameters adopted in this work. In Figure 4, we show the B850 and B800 FCE lineshapes along with the experiment at 300 K.¹⁰ While the overall broadening of the B850 band is reproduced, the asymmetry is not. The source of this asymmetry could be traced to the static disorder distribution and to the coupling of BChl states to charge-transfer states.⁵¹ On the other hand, the shape of the B800 band is almost perfectly reproduced by the FCE approach. In particular, the vibronic tail, which could not be reproduced by the mR theory, fits the experimental spectrum almost perfectly.

In Figure S4 (Supporting Information), we show the dissection of the LH2 lineshape into B850 and B800 contributions. The vibronic sidebands are clearly visible in the B800 spectrum and reflect in the total LH2 lineshape. In addition, there is a non-negligible contribution of B850 pigments also in the B800 region, which slightly alters the B800 lineshape in the full spectrum. To characterize the effect of the weak exciton coupling on the B800 lineshape, we compare in Figure S5 the B800 lineshape with the monomer lineshape obtained with the same spectral density and static disorder. The

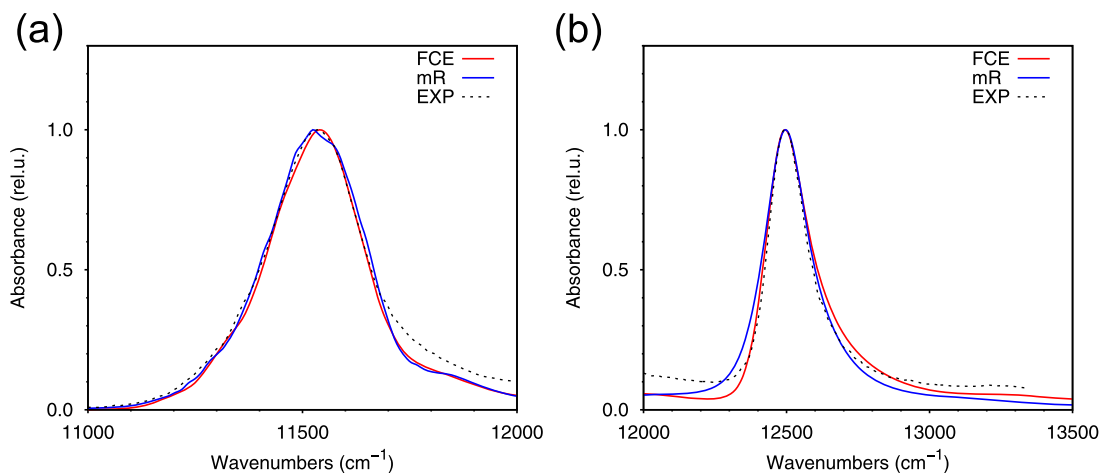


Figure 2. Absorption spectrum of LH2 at $T = 77$ K, obtained using FCE (red) and modified Redfield theory (blue) with the @CRY calculated exciton parameters (see text) and compared with the experiment at 77 K (dashed black line): (a) B850 band and (b) B800 band. Computed spectra are shifted to the experiment absorption maximum and normalized.

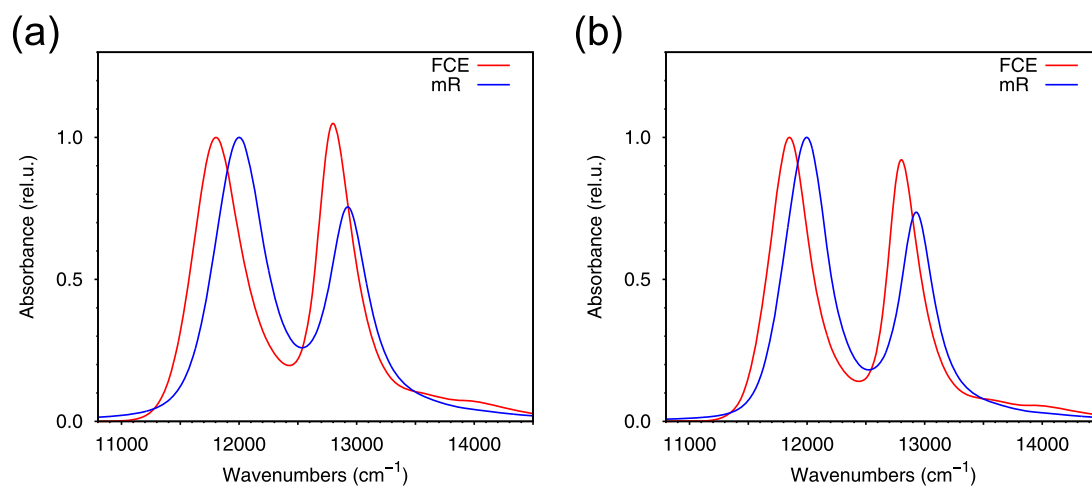


Figure 3. Absorption spectrum of LH2 at $T = 300$ K, obtained with the @MD calculated exciton parameters (see text) using FCE (red) and modified Redfield theory (blue). (a) Disorder parameters as in ref 13 and (b) disorder parameters slightly modified ($\sigma = 270$ cm^{-1} , $\lambda_c = 140$ cm^{-1} for B850, and $\lambda_c = 40$ cm^{-1} for B800).

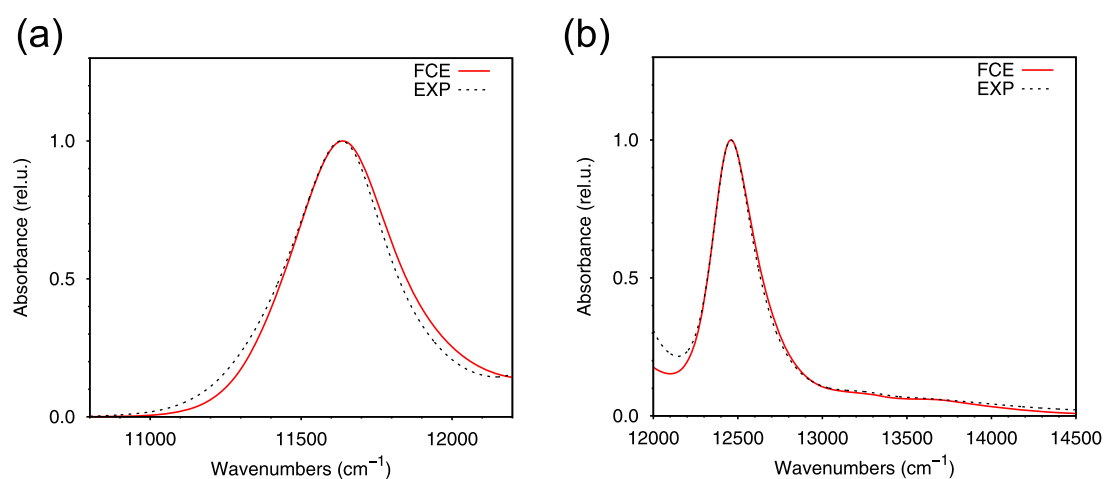


Figure 4. Absorption spectrum of LH2 at $T = 300$ K, obtained with the @MD calculated exciton parameters (see text) using FCE (red), compared with the experiment at 300 K (dashed black line). Left: B850 band; Right: B800 band. The disorder parameters are $\sigma = 270$ cm^{-1} , $\lambda_c = 140$ cm^{-1} for B850; $\sigma = 40$ cm^{-1} and $\lambda_c = 40$ cm^{-1} for B800. Computed spectra are shifted to the experiment absorption maximum and normalized.

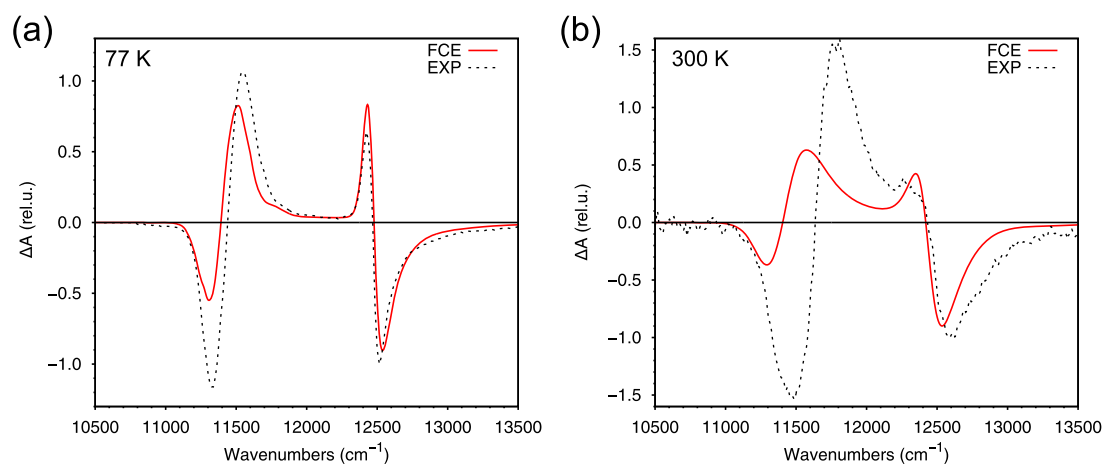


Figure 5. Circular dichroism spectra of LH2 computed using FCE (red) and compared with the experimental spectra (dashed black line): (a) spectrum at 77 K from the @CRY data; the computed spectrum was shifted by 1403 cm^{-1} to match the position of the B800 absorption and (b) spectrum at 300 K from the @MD data; the computed spectrum was shifted by -343 cm^{-1} to match the position of the B800 absorption. All spectra were normalized to one at the negative B800 peak.

exciton coupling gives rise to a small but noticeable red shift and to a more asymmetrical shape of the main absorption band. In addition, a small intensity redistribution from the vibronic tail to the main peak can be noticed.

3.2. Circular Dichroism Spectra. The CD spectra of LH2 computed at 77 K and at 300 K with FCE are reported in Figure 5, along with their experimental counterparts. Despite the excellent agreement in the absorption lineshapes with the experiment at both temperatures, the computed CD spectrum at 300 K deviates considerably from the experiments, especially in the B850 couplet. It was already observed in ref 10 that the intensity of the negative B850 band was underestimated by the calculations. Indeed, the sign and magnitude of the B850 couplet were shown to be extremely sensitive to small changes in the orientation of the transition dipole moments of the B850 BCHls.

Overall, the qualitative shape of the CD spectrum at 77 K is similar to that obtained in ref 10 using only Lorentzian broadening. On the contrary, in the 300 K spectrum computed here, the B800 couplet is somewhat narrower than the prediction from MD calculations in ref 10. It should be noted, however, that here we are employing a frozen arrangement of the chromophores, as calculated on the crystal structure, thus neglecting the fluctuations of couplings and transition dipoles, which affected the spectra predicted in ref 10.

4. DISCUSSION AND CONCLUSIONS

We have shown that the inclusion of non-Markovian effects and reorganization energy shifts through FCE theory has a considerable impact on the two exciton bands of LH2. The most striking differences are seen in the B800 band, for which the Redfield-based theories are known to have limitations arising from the near-degeneracy of exciton states.³ This effect is less pronounced in the B850 band due to the larger couplings and static disorder, which increase the separation between exciton energies. The reorganization energy shifts, on the other hand, can be easily taken into account also in Redfield-like theories, by including the imaginary part of the relaxation tensor.^{4,27}

The FCE approach yields an excellent agreement with the experiment for the B800 absorption, including its vibronic tail. This tail arises from the discrete part of the spectral density in eq 21, which in our case was calculated by means of multiscale QM methods.¹³ The spectral densities employed in this work present several peaks in the region at around 750 and 1200 cm^{-1} and over 1500 cm^{-1} , which contribute to the vibronic tail. The strong underestimation of this tail by mR theory can be traced back to the Markov treatment of all off-diagonal exciton–phonon interactions.^{20,23} In fact, the mR theory treats only the diagonal part of the exciton–phonon coupling in a time-dependent fashion, whereas the off-diagonal part is reduced to a simple lifetime symmetric broadening.

It has been suggested that nonsecular/non-Markov theories take into account localization effects, as they tend to reduce the exciton delocalization effects predicted by Markov/secular Redfield theory.²⁶ In particular, these localization effects are expected to reduce the intensity of the CD bands, as suggested in ref 26 for a chlorophyll dimer. To verify this hypothesis, we compared the CD spectrum calculated at room temperature by FCE and mR theories (Figure S2 in the Supporting Information). On the one hand, the B800 couplet is more intense for FCE than for mR theory, contrary to the expectations. We attribute this effect to the narrower overall shape predicted by FCE for the B800 band. On the other hand, a larger asymmetry in CD intensities is found for the mR lineshape

between the B850 and B800 couplets. The B850 band has an overall more positive CD, and the B800 band has a net negative CD. In the FCE lineshape, this asymmetry is reduced, which points to reduced delocalization between the B850 and B800 rings by including non-Markov and nonsecular terms.

We have computed the spectra of LH2 at different temperatures using distinct exciton Hamiltonians, which have been derived in ref 10 from crystal structure calculations (@CRY) and molecular dynamics calculations (@MD). Indeed, there is spectroscopic evidence that the nearest-neighbor couplings in the B850 ring reduce when increasing the temperature.^{47,52} We also assessed how the LH2 spectra change between 77 and 300 K without changing the exciton Hamiltonian. To this end, we computed the LH2 spectra at 77 and 300 K using only the @CRY exciton Hamiltonian (Figure S3a,b in the Supporting Information). By keeping the exciton Hamiltonian fixed, it is not possible to reproduce the blue shift of the B850 band, which is, however, apparent in the experiments (Figure S3c,d). Therefore, we conclude that two different exciton Hamiltonians are appropriate for low and room temperature.

Calculation of lineshapes in light-harvesting complexes has been an active research area. Theoretical methods are primarily based on second-order quantum master equations, including various versions of Redfield equations. In this context, the FCE formalism is the most basic second-order form, from which various Redfield equations can be derived with further approximations. For example, one subtle issue is the time order of the system–bath interaction operator, which is strictly imposed in the cumulant form adopted in FCE but approximately obeyed in the differential form adopted in TCL2 or TC2 quantum master equations.⁵³ In addition to accuracy, FCE also provides a convenient starting point to calculate emission spectra and the Forster energy transfer rate in molecular aggregates. However, we note that calculating emission spectra presents additional challenges related to the entanglement between the system and the bath created when the former relaxes in the excited state, assuming that a factorized initial condition leads to a qualitative difference from the exact results.²⁸ A comparison of several second-order approximations and FCE has been reported in ref 28 both analytically and numerically. This study further establishes its validity using the important light-harvesting system of LH2 and extends the FCE method to the calculation of CD spectra.

Correct treatment of the absorption lineshape is fundamental to assess the quality of a computed exciton Hamiltonian by comparing optical spectra with experiments. We have shown that, for LH2, the apparent frequency gap between the B800 and B850 bands is strongly influenced by the approximations introduced by the mR theory. The same effect might also appear in other exciton systems, where different exciton states have different degrees of localization. In the major and minor antenna complexes of higher plants, for example, there are pigment clusters in which excitation is delocalized over two or three chromophores and other more isolated pigments where the excitation is virtually localized.^{54–56} We envision that a more accurate lineshape theory such as the FCE will help refine the exciton Hamiltonian models of these antenna complexes.

From our results, it appears that the modified Redfield approach introduces a bias in the estimation of lineshape, which might reflect in an incorrect assessment of exciton Hamiltonian parameters when comparing optical spectra with experiments. Gelzinis et al. have reached similar conclusions by studying a

model dimer with different lineshape theories.²³ They noticed that Redfield-like approaches have been used to extract Hamiltonian parameters from experimental spectra and argued that these fitted parameters should be reassessed with care. Indeed, lineshapes have been customarily modeled with Redfield-like approaches, both for directly fitting site energies and couplings^{6,22,47,56} and for refining exciton Hamiltonians calculated by atomistic methods.⁵⁴

In conclusion, we have presented the implementation of the full cumulant expansion approach to computing the absorption and CD lineshapes of molecular aggregates, overcoming the usual Markovian and secular approximations made in the widely employed modified Redfield theory. Our implementation does not impose any restriction on the functional form of the spectral density, allowing efficient computation of linear spectra for systems with more than 20 states. As an example, we have shown the application of the FCE method to the LH2 antenna of purple bacteria, demonstrating that the FCE overcomes the limitations of the modified Redfield theory and better reproduces absorption lineshapes, including vibronic sidebands. The FCE approach will represent a valuable tool to achieve a quantitative comparison between calculated exciton Hamiltonians and experimental optical spectra.

■ ASSOCIATED CONTENT

SI Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.jpcb.0c05180>.

Detailed reorganization energy and disorder parameters; relevant parameters of the Hamiltonians (@CRY and @MD); plots of two sample realizations of the static disorder; comparison of CD spectra with mR and FCE; temperature dependence of the LH2 lineshape at fixed exciton Hamiltonian; the contribution of B800 and B850 rings to the LH2 lineshape; comparison between lineshapes of B800 and monomer BChl (PDF)

Complete exciton parameters (XLSX)

Numerical spectral densities (ZIP)

■ AUTHOR INFORMATION

Corresponding Author

Lorenzo Cupellini – Dipartimento di Chimica e Chimica Industriale, University of Pisa, S6124 Pisa, Italy; orcid.org/0000-0003-0848-2908; Email: lorenzo.cupellini@unipi.it

Authors

Filippo Lipparini – Dipartimento di Chimica e Chimica Industriale, University of Pisa, S6124 Pisa, Italy; orcid.org/0000-0002-4947-3912

Jianshu Cao – Department of Chemistry, Massachusetts Institute of Technology, Cambridge, Massachusetts 02139, United States; orcid.org/0000-0001-7616-7809

Complete contact information is available at: <https://pubs.acs.org/doi/10.1021/acs.jpcb.0c05180>

Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

We thank Prof. Benedetta Mennucci for fruitful discussions. L.C. acknowledges funding by the European Research Council, under the grant ERC-AdG-786714 (LIFETimeS). J.C. acknowl-

edges the support by the NSF (CHE 1800301 and CHE 1836913).

■ REFERENCES

- (1) Curutchet, C.; Mennucci, B. Quantum Chemical Studies of Light Harvesting. *Chem. Rev.* **2017**, *117*, 294–343.
- (2) Mirkovic, T.; Ostroumov, E. E.; Anna, J. M.; van Grondelle, R.; Govindjee; Scholes, G. D. Light Absorption and Energy Transfer in the Antenna Complexes of Photosynthetic Organisms. *Chem. Rev.* **2017**, *117*, 249–293.
- (3) Novoderezhkin, V. I.; van Grondelle, R. Spectra and Dynamics in the B800 Antenna: Comparing Hierarchical Equations, Redfield and Förster Theories. *J. Phys. Chem. B* **2013**, *117*, 11076–11090.
- (4) Renger, T.; Marcus, R. A. On the Relation of Protein Dynamics and Exciton Relaxation in Pigment-Protein Complexes: An Estimation of the Spectral Density and a Theory for the Calculation of Optical Spectra. *J. Chem. Phys.* **2002**, *116*, 9997–10019.
- (5) Georgakopoulou, S.; Frese, R. N.; Johnson, E.; Koolhaas, C.; Cogdell, R. J.; van Grondelle, R.; van der Zwan, G. Absorption and CD Spectroscopy and Modeling of Various LH2 Complexes from Purple Bacteria. *Biophys. J.* **2002**, *82*, 2184–2197.
- (6) Rancova, O.; Abramavicius, D. Static and Dynamic Disorder in Bacterial Light-Harvesting Complex LH2: A 2DES Simulation Study. *J. Phys. Chem. B* **2014**, *118*, 7533–7540.
- (7) Segatta, F.; Cupellini, L.; Garavelli, M.; Mennucci, B. Quantum Chemical Modeling of the Photoinduced Activity of Multichromophoric Biosystems. *Chem. Rev.* **2019**, *119*, 9361–9380.
- (8) Cupellini, L.; Bondanza, M.; Nottoli, M.; Mennucci, B. Successes & Challenges in the Atomistic Modeling of Light-Harvesting and its Photoregulation. *Biochim. Biophys. Acta, Bioenerg.* **1861**, 2020, No. 148049.
- (9) Jurinovich, S.; Viani, L.; Prandi, I. G.; Renger, T.; Mennucci, B. Towards an Ab Initio Description of the Optical Spectra of Light-Harvesting Antennae: Application to the CP29 Complex of Photosystem II. *Phys. Chem. Chem. Phys.* **2015**, *17*, 14405–16.
- (10) Cupellini, L.; Jurinovich, S.; Campetella, M.; Caprasecca, S.; Guido, C. A.; Kelly, S. M.; Gardiner, A. T.; Cogdell, R.; Mennucci, B. An Ab Initio Description of the Excitonic Properties of LH2 and Their Temperature Dependence. *J. Phys. Chem. B* **2016**, *120*, 11348–11359.
- (11) Lee, M. K.; Huo, P.; Coker, D. F. Semiclassical Path Integral Dynamics: Photosynthetic Energy Transfer with Realistic Environment Interactions. *Annu. Rev. Phys. Chem.* **2016**, *67*, 639–668.
- (12) Lee, M. K.; Bravaya, K. B.; Coker, D. F. First-Principles Models for Biological Light-Harvesting: Phycobiliprotein Complexes from Cryptophyte Algae. *J. Am. Chem. Soc.* **2017**, *139*, 7803–7814.
- (13) Segatta, F.; Cupellini, L.; Jurinovich, S.; Mukamel, S.; Dapor, M.; Taioli, S.; Garavelli, M.; Mennucci, B. A Quantum Chemical Interpretation of Two-Dimensional Electronic Spectroscopy of Light-Harvesting Complexes. *J. Am. Chem. Soc.* **2017**, *139*, 7558–7567.
- (14) Corbella, M.; Cupellini, L.; Lipparini, F.; Scholes, G. D.; Curutchet, C. Spectral Variability in Phycocyanin Cryptophyte Antenna Complexes is Controlled by Changes in the α -Polypeptide Chains. *ChemPhotoChem* **2019**, *3*, 945–956.
- (15) Mukamel, S. *Principles of Nonlinear Optical Spectroscopy*; Oxford University Press: New York, 1995.
- (16) Tanimura, Y.; Kubo, R. Time Evolution of a Quantum System in Contact with a Nearly Gaussian-Markoffian Noise Bath. *J. Phys. Soc. Jpn.* **1989**, *58*, 101–114.
- (17) Tanimura, Y. Stochastic Liouville, Langevin, Fokker–Planck, and Master Equation Approaches to Quantum Dissipative Systems. *J. Phys. Soc. Jpn.* **2006**, *75*, No. 082001.
- (18) Ishizaki, A.; Fleming, G. R. Unified Treatment of Quantum Coherent and Incoherent Hopping Dynamics in Electronic Energy Transfer: Reduced Hierarchy Equation Approach. *J. Chem. Phys.* **2009**, *130*, No. 234111.
- (19) Hein, B.; Kreisbeck, C.; Kramer, T.; Rodríguez, M. Modelling of Oscillations in Two-Dimensional Echo-Spectra of the Fenna–Matthews–Olson Complex. *New J. Phys.* **2012**, *14*, No. 023018.

- (20) Moix, J. M.; Ma, J.; Cao, J. Förster Resonance Energy Transfer, Absorption and Emission Spectra in Multichromophoric Systems. III. Exact Stochastic Path Integral Evaluation. *J. Chem. Phys.* **2015**, *142*, No. 094108.
- (21) Renger, T.; Müh, F. Understanding Photosynthetic Light-Harvesting: a Bottom up Theoretical Approach. *Phys. Chem. Chem. Phys.* **2013**, *15*, 3348.
- (22) Novoderezhkin, V. I.; van Grondelle, R. Physical Origins and Models of Energy Transfer in Photosynthetic Light-Harvesting. *Phys. Chem. Chem. Phys.* **2010**, *12*, 7352–7365.
- (23) Gelzinis, A.; Abramavicius, D.; Valkunas, L. Absorption Lineshapes of Molecular Aggregates Revisited. *J. Chem. Phys.* **2015**, *142*, No. 154107.
- (24) Jang, S.; Silbey, R. J. Theory of Single Molecule Line Shapes of Multichromophoric Macromolecules. *J. Chem. Phys.* **2003**, *118*, 9312–9323.
- (25) Schröder, M.; Kleinekathöfer, U.; Schreiber, M. Calculation of Absorption Spectra for Light-Harvesting Systems using non-Markovian Approaches as well as Modified Redfield Theory. *J. Chem. Phys.* **2006**, *124*, No. 084903.
- (26) Dinh, T.-C.; Renger, T. Towards an Exact Theory of Linear Absorbance and Circular Dichroism of Pigment-Protein Complexes: Importance of Non-Secular Contributions. *J. Chem. Phys.* **2015**, *142*, No. 034104.
- (27) Dinh, T.-C.; Renger, T. Lineshape Theory of Pigment-Protein Complexes: How the Finite Relaxation Time of Nuclei Influences the Exciton Relaxation-Induced Lifetime Broadening. *J. Chem. Phys.* **2016**, *145*, No. 034105.
- (28) Ma, J.; Cao, J. Förster Resonance Energy Transfer, Absorption and Emission Spectra in Multichromophoric Systems. I. Full Cumulant Expansions and System-Bath Entanglement. *J. Chem. Phys.* **2015**, *142*, No. 094106.
- (29) Ke, Y.; Zhao, Y. Hierarchy of Stochastic Schrödinger Equation Towards the Calculation of Absorption and Circular Dichroism Spectra. *J. Chem. Phys.* **2017**, *146*, No. 174105.
- (30) Chenu, A.; Cao, J. Construction of Multichromophoric Spectra from Monomer Data: Applications to Resonant Energy Transfer. *Phys. Rev. Lett.* **2017**, *118*, No. 013001.
- (31) Song, K.; Bai, S.; Shi, Q. A Time Domain Two-Particle Approximation to Calculate the Absorption and Circular Dichroism Line Shapes of Molecular Aggregates. *J. Chem. Phys.* **2015**, *143*, No. 064109.
- (32) Yang, M. A Reduced Density-Matrix Theory of Absorption Line Shape of Molecular Aggregate. *J. Chem. Phys.* **2005**, *123*, No. 124705.
- (33) Fetherolf, J. H.; Berkelbach, T. C. Linear and Nonlinear Spectroscopy from Quantum Master Equations. *J. Chem. Phys.* **2017**, *147*, No. 244109.
- (34) Cogdell, R. J.; Gall, A.; Köhler, J. The Architecture and Function of the Light-Harvesting Apparatus of Purple Bacteria: from Single Molecules to In Vivo Membranes. *Q. Rev. Biophys.* **2006**, *39*, 227–98.
- (35) Robert, B.; Cogdell, R. J.; van Grondelle, R. *Light-Harvesting Antennas in Photosynthesis*; Green, B. R.; Parson, W. W., Eds.; Springer: Netherlands, Dordrecht, 2003; pp 169–194.
- (36) Cupellini, L.; Lipparini, F. *FCEx Program to Compute Optical Spectra with the Full Cumulant Expansion*. OpenAIRE; 2020.
- (37) Jang, S.; Dempster, S. E.; Silbey, R. J. Characterization of the Static Disorder in the B850 Band of LH2⁺. *J. Phys. Chem. B* **2001**, *105*, 6655–6665.
- (38) Hyeon-Deuk, K.; Tanimura, Y.; Cho, M. Ultrafast Exciton-Exciton Coherent Transfer in Molecular Aggregates and its Application to Light-Harvesting Systems. *J. Chem. Phys.* **2007**, *127*, No. 075101.
- (39) Jang, S.; Silbey, R. J.; Kunz, R.; Hofmann, C.; Köhler, J. Is There Elliptic Distortion in the Light Harvesting Complex 2 of Purple Bacteria. *J. Phys. Chem. B* **2011**, *115*, 12947–12953.
- (40) Cleary, L.; Cao, J. Optimal Thermal Bath for Robust Excitation Energy Transfer in Disordered Light-Harvesting Complex 2 of Purple Bacteria. *New J. Phys.* **2013**, *15*, No. 125030.
- (41) Cleary, L.; Chen, H.; Chuang, C.; Silbey, R. J.; Cao, J. Optimal Fold Symmetry of LH2 Rings on a Photosynthetic Membrane. *Proc. Natl. Acad. Sci. U.S.A.* **2013**, *110*, 8537–8542.
- (42) Jang, S.; Rivera, E.; Montemayor, D. Molecular Level Design Principle Behind Optimal Sizes of Photosynthetic LH2 Complex: Taming Disorder Through Cooperation of Hydrogen Bonding and Quantum Delocalization. *J. Phys. Chem. Lett.* **2015**, *6*, 928–934.
- (43) De Vico, L.; Anda, A.; Osipov, V. A.; Madsen, AØ.; Hansen, T. Macrocycle Ring Deformation as the Secondary Design Principle for Light-Harvesting Complexes. *Proc. Natl. Acad. Sci. U.S.A.* **2018**, *115*, E9051–E9057.
- (44) Cardoso Ramos, F.; Nottoli, M.; Cupellini, L.; Mennucci, B. The Molecular Mechanisms of Light Adaption in Light-Harvesting Complexes of Purple Bacteria Revealed by a Multiscale Modeling. *Chem. Sci.* **2019**, *10*, 9650–9662.
- (45) Freiberg, A.; Timpmann, K.; Trinkunas, G. Spectral Fine-Tuning in Excitonically Coupled Cyclic Photosynthetic Antennas. *Chem. Phys. Lett.* **2010**, *500*, 111–115.
- (46) Kunz, R.; Timpmann, K.; Southall, J.; Cogdell, R. J.; Köhler, J.; Freiberg, A. Fluorescence-Excitation and Emission Spectra from LH2 Antenna Complexes of *Rhodospseudomonas acidophila* as a Function of the Sample Preparation Conditions. *J. Phys. Chem. B* **2013**, *117*, 12020–12029.
- (47) Pajusalu, M.; Rätsep, M.; Trinkunas, G.; Freiberg, A. Davydov Splitting of Excitons in Cyclic Bacteriochlorophyll a Nanoaggregates of Bacterial Light-Harvesting Complexes Between 4.5 and 263 K. *ChemPhysChem* **2011**, *12*, 634–44.
- (48) Kunz, R.; Timpmann, K.; Southall, J.; Cogdell, R. J.; Freiberg, A.; Köhler, J. Single-Molecule Spectroscopy Unmasks the Lowest Exciton State of the B850 Assembly in LH2 from *Rps. acidophila*. *Biophys. J.* **2014**, *106*, 2008–2016.
- (49) Harel, E.; Engel, G. S. Quantum Coherence Spectroscopy Reveals Complex Dynamics in Bacterial Light-Harvesting Complex 2 (LH2). *Proc. Natl. Acad. Sci. U.S.A.* **2012**, *109*, 706–711.
- (50) Ferretti, M.; Hendriks, R.; Romero, E.; Southall, J.; Cogdell, R. J.; Novoderezhkin, V. I.; Scholes, G. D.; van Grondelle, R. Dark States in the Light-Harvesting complex 2 Revealed by Two-dimensional Electronic Spectroscopy. *Sci. Rep.* **2016**, *6*, No. 20834.
- (51) Cupellini, L.; Caprasecca, S.; Guido, C. A.; Müh, F.; Renger, T.; Mennucci, B. Coupling to Charge Transfer States is the Key to Modulate the Optical Bands for Efficient Light Harvesting in Purple Bacteria. *J. Phys. Chem. Lett.* **2018**, *9*, 6892–6899.
- (52) Trinkunas, G.; Zerlauskienė, O.; Urbonienė, V.; Chmeliov, J.; Gall, A.; Robert, B.; Valkunas, L. Exciton Band Structure in Bacterial Peripheral Light-Harvesting Complexes. *J. Phys. Chem. B* **2012**, *116*, 5192–5198.
- (53) Cao, J. A Phase-Space Study of Bloch–Redfield Theory. *J. Chem. Phys.* **1997**, *107*, 3204–3209.
- (54) Müh, F.; Madjet, M. E.-A.; Renger, T. Structure-Based Identification of Energy Sinks in Plant Light-Harvesting Complex II. *J. Phys. Chem. B* **2010**, *114*, 13517–13535.
- (55) Müh, F.; Lindorfer, D.; Schmidt am Busch, M.; Renger, T. Towards a Structure-Based Exciton Hamiltonian for the CP29 Antenna of Photosystem II. *Phys. Chem. Chem. Phys.* **2014**, *16*, 11848–11863.
- (56) Novoderezhkin, V. I.; Palacios, M. A.; van Amerongen, H.; van Grondelle, R. Excitation Dynamics in the LHCII Complex of Higher Plants: Modeling Based on the 2.72 Å Crystal Structure. *J. Phys. Chem. B* **2005**, *109*, 10493–10504.